

VI Semester B.Sc. Degree (CBCSS – Reg./Supple./Imp.) Examination, May 2018 Core Course in Physics 6B13 PHY : QUANTUM MECHANICS (2014 Admn. Onwards)

Time: 3 Hours 18 Ve891 Salvoiene clienta mumbism peorly enough Max. Marks: 40

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Answer all – very short answer type – each question carries 1 mark.

- 1. The Wilson-Sommerfield quantization rule is ________
- 2. Write down the de Broglie relation. O.S. boa Ve Q. No asigners dilw anoxidate at
- The photoelectric effect establishes that light travels in the form of ____
- 4. Zeeman effect is the confirmation of ______ (1x4=4)

SECTION - B

Answer any seven – short answer type – each question carries two marks.

- List out the basic experimental results of the photoelectric phenomena.
- Explain the assumptions of Planck with regard to cavity radiation.
- 7. Illustrate the uncertainty principle on the basis of single slit experiment.
- 8. What are stationary states?
- 9. Outline the various admissibility conditions on the wavefunction of a system.
- 10. What is meant by expectation value of a dynamical variable?
- 11. A particle confined in a box must have a certain minimum energy called zero point energy. Comment. O.T.q Obtain Schrödinger equation for the hydrogen atom in spherical p



- 12. Explain Zeeman effect.
- 13. Explain the magnetic quantum number of an atom.
- 14. What does tunnelling mean?

(2×7=14)

SECTION - C

Answer any four - short essay/problem - each question carries three marks.

- 15. From a sodium surface, light of wavelength 3125 Å and 3650 Å causes emission of electrons whose maximum kinetic energy is 2.128eV and 1.595 eV, respectively. Estimate Planck's constant and work function of sodium.
- 16. The average lifetime of an excited atomic state is 10-9s. If the spectral line associated with the delay of this state is 6000 Å, estimate the width of the line.
- 17. Explain how barrier tunnelling accounts for α decay by certain nuclei.
- Electrons with energies of 1.0 eV and 2.0 eV are incident on a barrier 10.0 eV high and 0.50 nm wide.
 - a) Find their respective transmission probabilities.
 - b) How are these affected if the barrier is doubled in width?
- 19. Verify that the average value of 1/r for a 1s electron in the hydrogen atom is $1/a_0$. Given $\psi = \frac{1}{\sqrt{\pi a_0^3}} e^{-r/a_0}$.
- 20. Discuss Stern-Gerlach experiment.

 $(3 \times 4 = 12)$

 $(5 \times 2 = 10)$

SECTION - D

Answer any two - long essay type - each question carries five marks.

- 21. What is Compton effect ? How does Compton effect provide a conclusive evidence of the particle properties of radiation ?
- 22. State and explain the postulates of quantum mechanics.
- 23. A particle is trapped in a square well potential of finite depth. Show that the particles have a nonzero probability of being found outside the well even if its energy is less than the height of the barriers.
- Obtain Schrödinger equation for the hydrogen atom in spherical polar coordinates.