

Reg. No. :	
Name .	2012003

III Semester M.Sc. Degree (CBSS - Reg./Suppl./Imp.) Examination, October 2021 (2018 Admission Onwards) CHEMISTRY

CHE3E.03: Polymers and Material Chemistry

Time: 3 Hours

Max. Marks: 60

SECTION - A

Answer all questions in one word or in one sentence. Each question carries one mark.

- 1. Write two examples for natural polymers.
- Write down the structures of polypropylene and polystyrene.
- 3. Write an empirical equation connecting intrinsic viscosity with molecular size.
- 4. What is a homogeneous polymer?
- Name two high energy radiations which cause polymer degradation.
- Mention the name of a polymer synthesised through solid phase polymerisation. Name ore of tantalum and mention its use.
- 8. What is porous bearing?

 $(8 \times 1 = 8)$

SECTION - B

Answer any eight questions. Answer may be in two or three sentences. Each question carries two marks.

- 9. What are thermoplastics? Give two examples.
- 10. Write the rate equation for the free-radical chain polymerization and explain the terms.

P.T.O.

11. What is meant by living polymer?

three marks.

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-2-



 $(8 \times 2 = 16)$

- 12. Mention Colligative properties that are used to measure the molecular mass of
- polymers. 13. How swelling occurs in a polymer?
- 14. What is meant by fractionation of polymers? Mention two methods used for the fractionation of polymers.
- heterogeneous polymerisation. 16. What is a Block Copolymer? Schematically represent block copolymers.

15. What is heterogeneous polymerization? Mention two methods used for the

- 17. What are biodegradable polymers? Mention the name of a biodegradable polymer.
- 18. What are Die Steels? Mention one of its use. 19. Write about four mechanical properties required for engineering materials.

20. What are the different optical properties to be considered for engineering

materials? SECTION - C Short paragraph questions. Answer any four questions. Each question carries

21. State and explain the organic and inorganic polymers. Give two examples for each. 22. Explain the mechanism of ring opening polymerization with a suitable example.

23. Briefly discuss the viscosity measurement for the determination of molecular size.

24. Explain the dissolution of polymer molecules in solution and schematically

represent the 'micellar colloid' of soap molecule and 'molecular colloid' of polymer molecule.

26. What are refractory materials? Explain the properties and applications of refractory materials with specific examples.

Catalysis.

materials.

solution polymerisation.

B) What is Ziegler-Natta Catalyst ? Explain the mechanism of Ziegler-Natta

B) Explain vapour pressure lowering method to measure the molecular weight

 $(4 \times 3 = 12)$

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SECTION - D Essay type questions. Answer four questions. Each question carries six marks. 27. A) Define condensation polymerization and explain different types of

condensation polymerization reactions with suitable examples.

25. What is suspension polymerisation? Mention two monomers which can be

used for solution polymerisation. Write about advantage and disadvantages of

- 28. A) Explain the use, principle and procedure of GPC. OR
- of polymer. 29. A) Explain the factors affecting thermal and mechanical, degradation of

- polymers. OR
- B) What is a graft copolymer? Write two types of graft copolymerisation reactions with suitable examples.

30. A) Explain the following (a) Bearing (b) Hybrid composite and (c) Ceramic

OR

B) Write short notes on (a) electrical properties and (b) magnetic properties of $(4 \times 6 = 24)$ engineering materials.