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Reg. No. :

Name :

Sixth Semester B.Sc. Degree (C.B.C.S.S. – Supplementary/ One Time

Mercy Chance) Examination, April 2024

(2014 to 2018 Admissions)

CORE COURSE IN PHYSICS

6B 13 PHY: Quantum Mechanics

Time: 3 Hours

Max. Marks: 40

Instruction: Write answers in English only.

SECTION - A

Answer all. Very short answer type. Each question carries one mark.

- Write the expression for energy of a harmonic oscillator.
- 2. Write down the expression for energy operator.
- Write down the time independent Schrodinger equation.
- 4. What is Compton effect?

 $(4 \times 1 = 4)$

SECTION - B

Answer any seven. Short answer type. Each question carries two marks.

- 5. What is the significance of Franck Hertz Experiment?
- Write the position-momentum uncertainty relation and the energy-time uncertainty relation.
- 7. What do you mean by eigen values and eigen functions?
- 8. What are the conditions that a wave function must satisfy to be acceptable?
- 9. What are free particles? Write its box normalized wave function.
- Explain the postulates of Bohr with regard to Hydrogen atom.

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- Using the idea of standing wave of an electron in the orbit, obtain Bohr's quantization rule.
- 12. Write any two quantum numbers and their permissible values.
- 13. What do you mean by tunnel effect ?
- 14. State and explain Exclusion principle.

 $(7 \times 2 = 14)$

SECTION - C

Answer any four. Short essay/problem type. Each question carries three marks.

- 15. Explain the phenomenon of Photoelectric effect.
- Determine the de Broglie wavelength of an electron that has been accelerated through a potential difference of 100 V.
- 17. When two observables are said to be compatible? State the theorems that indicate the connection between compatible observables and commuting operators.
- 18. Explain briefly the Zeeman effect.
- 19. A 1 eV electron got trapped inside the surface of a metal. If the potential barrier is 4.0 eV and the width of the barrier is 2Å, calculate the probability of its transmission.
- Using the rule of quantization, obtain the expression for energy of a particle in a box. (4x3=12)

SECTION - D

Answer any two. Long essay type. Each question carries five marks.

- Obtain the expressions for lengths of semi major axis of nth orbit of an electron orbit of Hydrogen atom.
- Write down the Schrodinger equation of a linear harmonic oscillator and obtain its energy eigen values.
- 23. Derive the Schrodinger equation of the Hydrogen atom in spherical polar coordinates. Write the equations for R, Θ and Φ using separation of variables.
- State and explain the postulates of quantum mechanics.

(2×5=10)