

Reg.	No.	:	
Mana			

V Semester B.Sc. Degree (CBCSS - OBE-Regular/Supplementary/ Improvement) Examination, November 2022

### (2019 Admission Onwards) CORE COURSE IN CHEMISTRY/POLYMER CHEMISTRY 5B08CHE/PCH : Inorganic Chemistry

Time: 3 Hours

Max. Marks: 40

### SECTION - A

Answer all questions. Each carries 1 mark.

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- 1. Basicity of hydroxides of lanthanides atomic number.
- Give one example of ambidentate ligand.
- The number of unpaired electrons in [Mn(H<sub>2</sub>O)<sub>6</sub>]<sup>2+</sup> is
- 4. Toxicity of lead is due to its capacity to mimic

#### SECTION - B

Answer any seven questions out of 10. Each carries 2 marks.

- Calculate the number of metal-metal bond in Fe<sub>3</sub>(CO)<sub>12</sub>.
- 6. Give the basic difference between graphene and graphite.
- Name two zinc containing enzymes. 8. Calculate the CFSE for a d4 system in tetrahedral field.
- 9. What is meant by chelate effect ?
- Give any two consequences of lanthanide contraction.
- 11. Explain why Cu2+ is more stable than Cu+ in aqueous solution.
- Calculate the EAN of Fe in [Fe(CN)<sub>6</sub>]<sup>4-</sup>.
- 13. Tetrahedral complexes are generally high-spin. Why?
- 14. What is Bohr effect?

 $(7 \times 2 = 14)$ 

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## SECTION - C

Answer any four questions out of 6. Each carries 3 marks.

- 15. Explain why transition metal compounds are coloured.
- Discuss the factors affecting the stability of complexes.
- 17. Discuss the structure of [Ni(CN)<sub>4</sub>]<sup>2-</sup> using Valene bond theory.
- 18. Write a note on biological nitrogen fixation.
- 19. Give the classification of organometallic compounds based on the nature of
- 20. Give the biological functions of Ca.

 $(4 \times 3 = 12)$ 

# SECTION - D

Answer any two questions out of 4. Each carries 5 marks.

- 21. Give the mechanism of sodium-potassium pump.
- 22. Discuss the preparation, properties and reactions of ferrocene.
- 23. Discuss the crystal field splitting in octahedral complexes.
- 24. Explain the separation of lanthanides by ion-exchange method.

 $(2 \times 5 = 10)$