



Reg. No. :

Name :

Third Semester B.Sc. Degree (CBCSS – Supplementary)
Examination, November 2022
(2016 – 18 Admissions)
COMPLEMENTARY COURSE IN CHEMISTRY
3C03CHE(PS) Chemistry (For Physical Science)

Time : 3 Hours

Max. Marks : 32

SECTION – A

Answer **all** questions. **Each** question carries 1 mark.

1. Give one example of an ambidentate ligand.
2. What is meant by homolysis ?
3. Name any two ores of aluminium.
4. What are closed systems ?
5. What is the hybridization and geometry of $[\text{Ni}(\text{CN})_4]^{2-}$? (1×5=5)

SECTION – B

Answer **any four** questions. **Each** question carries 2 marks.

6. What is n/p ratio ? How is it related to stability ?
7. What is the physical significance of Gibbs free energy ?
8. What are Grignard reagents ? Give one method for preparation.
9. What is froth flotation process ?
10. What is meant by zero point energy ?
11. What is electromeric effect ? (2×4=8)

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SECTION – C

Answer **any three** questions. **Each** question carries 3 marks.

12. State and formulate the first law of thermodynamics. Calculate the internal energy change produced when 800 J of work is done on a system which gives off 220 J of heat.
13. Give an account of the problems associated with nuclear waste disposal.
14. Give the mechanism of Friedel Crafts acylation and alkylation reactions.
15. What are the limitations of VB theory ?
16. Write notes on :
 - a) ionic and
 - b) pi bonded organometallic compounds. (3×3=9)

SECTION – D

Answer **any two** questions. **Each** question carries 5 marks.

17. a) Explain radiocarbon dating.
b) An old wooden furniture shows a C^{14} activity which is 60% of the activity of fresh wood. Find the age of the wood that was used to make the furniture. Half life of C^{14} is 5760 years.
18. Explain the various stages in the metallurgy of Ni.
19. Discuss the reactions of ferrocene.
20. Write notes on :
 - a) spin-spin coupling
 - b) finger print region
 - c) chemical shift. (5×2=10)