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Reg. No.:....

Third Semester B.Sc. Degree (CBCSS – OBE – Regular/Supplementary/
Improvement) Examination, November 2022
(2019 Admission Onwards)

CORE COURSE IN CHEMISTRY/POLYMER CHEMISTRY
3B04CHE/PCH : Organic Chemistry – I

Time: 3 Hours

Max. Marks: 40

Instruction: Answer the questions in English only.

SECTION - A

Very short answer type. Each carries 1 mark. Answer all 4 questions.

- When benzene is treated with n-propyl chloride in presence of AlCl₃, the product is
- 2. Give an example for edible dye.
- Give an example for a compound showing optical isomerism without any stereocentre.
- 4. What is the hybridization of carbon in triplet carbene?

 $(4 \times 1 = 4)$

SECTION - B

Short answer type. Each carries 2 marks. Answer any 7 questions out of 10.

- Arrange the following in the increasing order of their Ka values. Justify your answer Fluoroacetic acid, Chloroacetic acid, Bromoacetic acid, Iodoacetic acid.
- 6. Differentiate between asymmetric and dissymmetric molecules.
- 7. What is meant by Ziegler Natta polymerization of alkenes?
- 8. Distinguish between chromophore and auxochrome. Give examples.

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- 9. How methyl orange is synthesized?
- 10. What are active methylene compounds? Give two examples.
- 11. What is the difference between conformation and configuration?
- 12. What are antiaromatic compounds? Give two examples.
- 13. How isotope effects are used for the determination of reaction mechanisms?
- 14. Which will undergo nitration more readily: Toluene or benzene? Why? (7×2=14)

SECTION - C

Short essay/problem type. Each carries 3 marks. Answer any 4 questions out of 6.

- a) Why HCOOH can be esterified easily than trimethyl acetic acid?b) Why NF₃ is a much weaker base than (CH₃)₃N?
- 16. Define specific rotation. Find out the specific rotation at 200C for sodium (D line) light for a solution of 3 g of an enantiomer dissolved in ethanol to make 100 mL of solution. The observed rotation is + 2.10° in 10 cm polarimeter tube.
- 17. What is Wittig reagent ? Give the preparation and synthetic application of it.
- 18. Explain the number average molecular weight and weight average molecular weight of a polymer.
- 19. Designate E, Z notation to each of the following.

Explain the following statement: - NO₂ group is a deactivating and meta directing whereas - CI group is a deactivating o, p directing towards electrophilic substitution. (4x3=12)

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SECTION - D

Long essay/problem type. Each carries 5 marks. Answer any 2 questions out of 4.

- 21. What are free radicals? Give an account of their structure and stability.
- 22. a) Why carboxylic acid is stronger acid than phenol?
 - b) Write a note on Reformatsky reaction.

relative

(2+3)

(3+2)

- a) Discuss the conformational analysis of n-butane and compare the relative stabilities of its various conformer.
 - b) Differentiate between racemization and resolution.

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- 24. a) How benzene can be converted into benzene Sulphonic acid? Explain with mechanism.
 - b) What are conducting polymers? Give two examples.

(2×5=10)

(3+2)