Reg. No.:

V Semester B.Sc. Degree (CBCSS – O.B.E. – Regular/Supplementary/ Improvement) Examination, November 2024 (2019 to 2022 Admissions) CORE COURSE IN CHEMISTRY/POLYMER CHEMISTRY 5B08CHE/PCH: Inorganic Chemistry

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Time: 3 Hours

Max. Marks: 40

SECTION - A

Answer all questions. Each question carries 1 mark.

 $(4 \times 1 = 4)$

- What is the common oxidation state of actinides ?
- 2. What are organometallic compounds?
- 3. Which macromolecule serves as the oxygen storage in muscle cells ?
- What is the hybridization of Ni in [Ni(CN)₄]²⁻?

SECTION - B

Answer any 7 questions. Each question carries 2 marks.

 $(7 \times 2 = 14)$

- Calculate the spin only magnetic moment of [Fe(CN)₆]³⁻.
- Explain the toxicity of cadmium.
- 7. Explain why tetrahedral complexes are unable to exhibit geometrical isomerism.
- Which is more basic, La(OH)₃ or Lu(OH)₃? Why?
- 9. What is chelation?
- 10. What is meant by crystal field splitting? What is the CFSE of a high spin octahedral complex with d⁵ configuration?

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- 11. Draw the structure of Fe₃(CO)₁₂.
- 12. Among $[Fe(CN)_6]^{3-}$ and $[Fe(CN)_6]^{4-}$, which has higher stability constant and why?
- 13. What are MXenes ? Give an example.
- 14. Explain why zinc complexes are generally colourless?

SECTION - C

Answer any 4 questions. Each question carries 3 marks.

 $(4 \times 3 = 12)$

- 15. Explain low spin and high spin complexes with suitable examples.
- 16. Write a short note on the occurrence of lanthanides.
- 17. What is lanthanide contraction? Explain its causes and consequences.
- 18. Discuss the structural features and functions of myoglobin.
- Write a short note on graphene.
- 20. Write a short note on EAN and its significance.

SECTION - D

Answer any 2 questions. Each question carries 5 marks.

 $(2 \times 5 = 10)$

- 21. Explain the role of hemoglobin in the respiration process.
- Discuss the structural isomerism exhibited by co-ordination compounds with suitable examples.
- 23. Discuss about the general characteristics of actinides.
- Explain how CFT describes magnetic properties and colour of co-ordination compounds.