

Reg. No. : ..... Name : .....



K20U 1496

V Semester B.Sc. Degree (CBCSS - Reg./Sup./Imp.) Examination, November 2020 (2014 Admn. Onwards) Core Course in Chemistry 5B08 CHE: INORGANIC CHEMISTRY - II

Time: 3 Hours

Max. Marks: 40

## SECTION - A

Answer all questions. Each question carries one mark.

- 1. What is EAN rule?
- What is the spin only magnetic moment value of [Fe(CN)<sub>B</sub>]<sup>3-</sup>?
- Give general electronic configuration of actinides.
- 4. List any two uses of non ferrous alloys. (4×1=4)

## SECTION - B

Answer any seven questions. Each question carries 2 marks.

- 5. What is meant by spectrochemical series?
- What are metalloenzymes? Name two Zn enzymes.
- 7. What is the difference between active transport and passive transport?
- Explain why Δt is smaller than Δο.
- What are chelating ligands? Give examples.
- 10. Compare the electronic spectra of lanthanides and transition metals.
- 11. What are the different methods for heat treatment of steel?
- 12. What is dry corrosion?
- 13. What are coinage metals? How do they occur in nature?
- 14. Distinguish between double salt and complex salt.

 $(7 \times 2 = 14)$ 



## SECTION - C

Answer any 4 questions. Each question carries 3 marks.

- 15. Explain the electrochemical theory of corrosion.
- Co(NH<sub>3</sub>)<sub>6</sub> is diamagnetic while CoF<sub>6</sub> is paramagnetic. Explain on the basis of CFT.
- 17. Write a note on geometrical isomerism of complexes.
- 18. What are the different ore concentration methods?
- 19. Give the structure and functions of Hb.
- 20. Explain the ion exchange method for separation lanthanides.

 $(4 \times 3 = 12)$ 

## SECTION - D

Answer any 2 questions. Each question carries 5 marks.

- 21. Discuss the causes and consequences of lanthanide contraction.
- Explain the splitting of d orbitals in tetrahedral and square planar ligand fields.
- What are the factors influencing corrosion? Discuss the methods to control corrosion.
- 24. Explain the biological functions and toxic effects of
  a) Co
  b) Ni
  c) Cu
  (2x5=10)