0092180 K19U 2460

III Semester B.Sc. Degree (CBCSS-Reg./Sup./Imp.) Examination, November - 2019 (2014 Admn. Onwards)

COMPLEMENTARY COURSE IN CHEMISTRY 3C03 CHE(BS): CHEMISTRY (FOR BIOLOGICAL SCIENCES)

Time: 3 Hours

Max. Marks: 32

SECTION-A

Answer All questions. Each question carries 1 mark.

 $(5 \times 1 = 5)$

- 1. What is meant by isolated systems?
- Define the term conformation. 2.
- Name the functional group in 3.
 - a) Carboxylic acid
 - b) Amide
- 4. What is meant by chirality ?
- 5. What are chelating ligands?

SECTION-B

Answer any Four questions. Each question carries 2 marks.

- 6. Name the following
 - a) Zn₂[Fe(CN)₆]
 - b) [Cr(NH₂)₆] Cl₃
- Predict the products CH₃ CH = CH₂ + HBr → 7.
- What is meant by heterolysis? Give one example.
- What are free radicals? Give any two reaction in which they are formed.
- 10. What are isochoric and isobaric processes?

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11. The boiling point of diethyl ether is 35°C. Its heat of vaporization at its boiling point is 27.2 KJ/mole. Calculate entropy of vaporization.

SECTION-C

Answer any Three questions. Each question carries 3 marks. (3x3=9)

- 12. Write down Gibbs Helmholtz equation. What are the criterion for spontaneity?
- 13. Discuss Werners theory of coordination.
- 14. Discuss the optical isomerism of tartaric acid.
- 15. Explain peroxide effect with a suitable example.
- 16. Give an account of formaldehyde based plastics.

SECTION-D

Answer any Two questions. Each question carries 5 marks. (2x5=10)

- 17. Discuss the factors affecting stability of complexes.
- 18. Explain with mechanism the various electrophilic substitution reactions of benzene.

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- a) Derive a relation between Cp and Cv.
 - b) State and explain second law of thermodynamics.
- 20. Write notes on
 - a) co polymers
 - b) biodegradable polymers.