



K16U 2093

Reg. No. : .....

Name : .....



III Semester B.Sc. Degree (CBCSS – Reg./Supple./Imp.)  
Examination, November 2016  
(2014 Admission Onwards)  
COMPLEMENTARY COURSE IN CHEMISTRY  
(For Physical Sciences) 3C03CHE(PS) : Chemistry

Time : 3 Hours

Max. Marks : 32

## SECTION – A

Answer all questions. Each question carries 1 mark.

1. What are isotones ? Give example.
2. Give one example of a poly dentate ligand.
3. Name any two ores of titanium.
4. What is meant by homologous series ?
5. What are closed and isolated systems ?

(1×5=5 Marks)

## SECTION – B

Answer any four questions. Each question carries 2 marks.

6. What are the applications of IR spectroscopy ?
7. Write a note on ionic organometallic compounds.
8. What is meant by N/P ratio ? How is it related to stability ?
9. State and explain second law of thermodynamics.
10. What is meant by EAN ? Explain giving examples.
11. Explain the aromaticity of benzene on the basis of Huckels rule.

(2×4=8 Marks)

P.T.O.

## SECTION - C

Answer **any three** questions. **Each** question carries **3** marks.

12. What are spontaneous and non spontaneous processes ? What are the criterion for spontaneity ?
13. Explain any two electron displacement effects.
14. Explain the following :
  - a) spin spin splitting
  - b) zero point energy
15. Explain how the complex formation is useful in qualitative analysis ?
16. A wooden sample shows a C-14 activity of 3.2 disintegration per minute per gram of carbon. A freshly cut wood shows an activity of 15.2 disintegration per minute per gram of carbon. Calculate the age the sample  
(Half life C-14 = 5770 years). (3×3=9 Marks)

## SECTION - D

Answer **any two** questions. **Each** question carries **5** marks.

17. a) What are the methods for separation of isotopes ? 3  
b) Discuss the problems associated with nuclear waste disposal. 2
  18. Give an outline of extractive metallurgy of Ni.
  19. Explain the bonding, structure and reactions of ferrcene.
  20. Explain the factors affecting the stability of complexes. (5×2=10 Marks)
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