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Reg. No.:....

Name :



K19U 3172

I Semester B.Sc. Degree (CBCSS-Supplementary / Improvement)
Examination, November - 2019
(2014-2018 Admissions)

COMPLEMENTARY COURSE IN CHEMISTRY

1C01 CHE: CHEMISTRY (FOR PHYSICAL AND BIOLOGICAL SCIENCE)

Time: 3 Hours

Max. Marks: 32

SECTION - A

Answer all questions. Each question carries 1 mark.

(5x1=5)

- 1. Write down de Broglie equation and define the terms.
- 2. What is meant by electrolysis?
- 3. Define transport number of an ion.
- 4. What is an ionic bond? Give one example for an ionic compound.
- 5. Why alkali metals have the lowest ionization energy?

SECTION - B

(Answer any Four questions. Each question carries 2 marks) (4x2=8)

- 6. State Heisenberg's uncertainty principle.
- 7. Distinguish between electro negativity and electron affinity.
- 8. State faraday's first and second laws of electrolysis.
- 9. Explain coagulation method for water purification.
- 10. What is the cause of lanthanide contraction?
- Arrive at the shape and bond angles of SF₆ molecule using VSEPR theory.



SECTION C

(Answer any Three questions. Each question carries 3 marks) (3x3=9)

- 12. Write a note on the three water quality parameters; DO, BOD and COD.
- Define lattice energy of ionic compounds. Suggest a method to determine lattice energy.
- 14. Difference between biological magnification and bioaccumulation.
- 15. How will you explain the geometry of NH₃ and H₂0 on the basis of hybridisation?
- 16. Explain molecular orbital theory for homonuclear diatomic molecules.

SECTION D

(Answer any Two questions. Each question carries 5 marks) (2x5=10)

- 17. What are the causes of air pollution? Mention their effect on biotic components.
- 18. Define specific and molar conductance. Explain their variation with dilution.
- 19. Discuss the atomic spectra of hydrogen atom. Calculate the wavelength limits of the Balmer line of Hydrogen atom.
- 20. What is meant by hydrogen bonding? Discuss different types of hydrogen bonding with suitable examples.