



M 26099

Reg. No. :

Name :

Third Semester M.A./M.Sc./M.Com. Degree (Reg./Sup./Imp.)
Examination, November 2014
(2009 Admn. Onwards)
PHYSICS
PH 303 : Nuclear Physics

Time: 3 Hours

Max. Marks : 50

SECTION – A

Answer **any two** questions. **Each** question carries **10** marks.

1. Using a square Nuclear potential well for a deuteron derive the relationship between the range of nuclear force and depth of potential well.
2. Using Fermi's theory of β decay deduce and expression for the probability of β -particle emission.
3. Explain compound nucleus reactions. List and explain the different conservation laws in nuclear reaction. Define Q value and threshold in nuclear reaction. Obtain their quantities in nuclear masses.
4. Explain diffusion of thermal neutrons from a plane surface. Derive an expression for the diffusion length. (2×10=20)

SECTION – B

Answer **any five** questions. **Each** question carries **3** marks.

5. What are magic numbers ? What is the magicity about magic numbers ?
6. What are stripping reactions ? What are pick-up reactions ?
7. Explain the basic principle of the working of a GM counter.
8. Why is it that alpha decay cannot be explained on classical grounds ?

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M 26099



9. What is reactor buckling ?
10. Give the statistical model of fission.
11. Explain internal conversion. What is an isomer ?
12. What is nuclear fusion ? What are the major hindrances in the design of controlled fusion reactor ?

(5×3=15)

SECTION – C

Answer **any three** questions. **Each** question carries **5** marks.

13. Discuss the Meson theory of nuclear forces. Define Yukawa potential and its variation with distance.
14. Explain the basic nuclear properties. Calculate the density of ${}_6\text{C}^{12}$ nucleus given radius of carbon nucleus = 2.7×10^{-15} m.
15. Explain the construction and working of a Scintillation counter.
16. What are the factors that contribute to the semi empirical mass formula ? Using the semi-empirical mass formula find the most stable isobar for a nucleus having odd A.
17. Discuss the essential difference between the scattering by neutron-proton system and proton-proton system.

(3×5=15)